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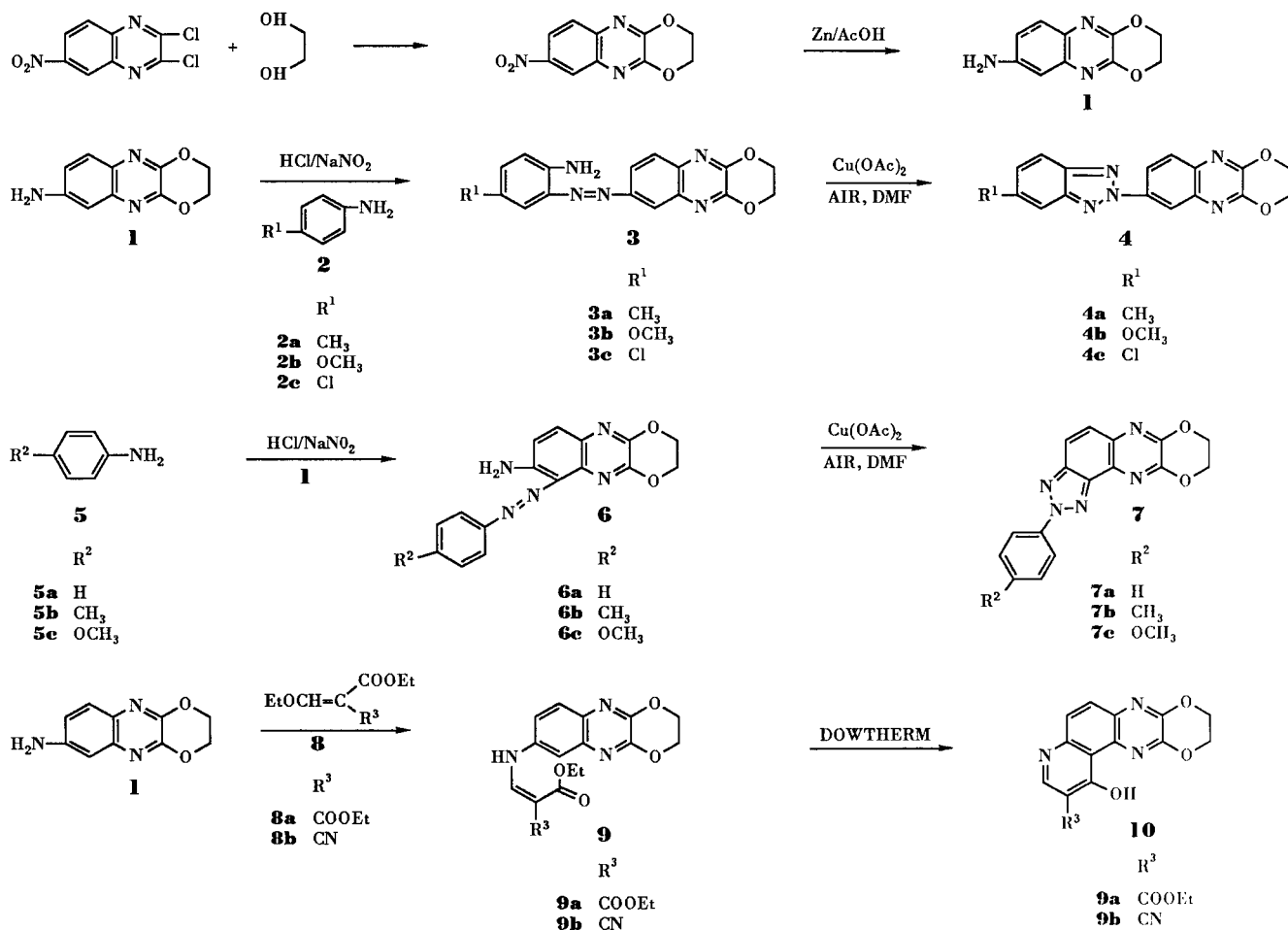
A novel efficient synthesis of fluorescent, fused quinoxalines was achieved. 7-Triazolyl-1,4-dioxino[2,3-*b*]quinoxalines were synthesized by the diazotisation of 7-amino-1,4-dioxino[2,3-*b*]quinoxaline and coupling with selected aromatic amines followed by air oxidation. Diazotised aryl amines were coupled with 7-amino-1,4-dioxino[2,3-*b*]quinoxalines followed by subsequent air oxidation afforded 1,4-dioxino[2,3-*b*]quinoxalino[6,5-*d*]1,2,3-triazoles. 7-Amino-1,4-dioxino[2,3-*b*]quinoxaline was condensed with conjugated enol ethers followed by cyclization in dowtherm resulted in 1,4-dioxino[2,3-*b*]quinoxalino[6,5-*b*]pyridines.

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Quinoxalines are commercially important as antibiotics [1], antagonists [2], agrochemicals [3], fungicides [4], herbicides [5]. Several patents describe the synthesis and technical importance of quinoxalines as hair dyes [6], cyanine dyes [7], reactive dyes [8], azo dyes [9], fluorescent dyes [10] and pigments [11]. However, in general there has been little exploitation of quinoxaline derivatives in the field of dyestuffs. We have recently reported the synthesis of novel heterocyclic dyes and fluorescent brighteners such as thiophenes [12-13], benzo[*b*]thiophenes [14],

thiazoles [15] and their applications on synthetic fibres. The versatility of quinoxalines in the dyestuff field [16-21] was also demonstrated by us.

The results of this study have encouraged us to explore the utility of compound **1** in the synthesis of fused and pendant heterocyclic fluorescent compounds. A fluorophoric heterocycle such as 1,2,3-triazole pendant to another heterocycle in a suitable position finds an exceedingly important place in commercial fluorescent whiteners illustrated in the patent literature [22-23]. Com-



pounds with 1,2,3-triazole attract special attention on account of their strong fluorescence [24]. Some fused and pendant 1,2,3-triazolyl compounds have been reported by us previously [16,25-26]. The condensation of a variety of aromatic and heterocyclic amino compounds with conjugated enol ethers followed by cyclization of the α -aminoethylene- β -carboxylates resulted in useful fused pyridine derivatives [16,27-28].

In this communication, we wish to report a facile synthesis of few hitherto unknown condensed quinoxalines by a novel method. 7-Amino-1,4-dioxino[2,3-*b*]quinoxaline **1** is a versatile intermediate in the synthesis of heterocyclic systems. It is interesting to study various characteristics reactions of **1** in developing novel, fluorescent heterocycles. The key compound **1** has been synthesized by the cyclocondensation reaction of 2,3-dichloro-6-nitroquinoxaline with 1,2-ethanediol followed by reduction [17].

In connection with our interest to study fluorescent properties of **4**, we have devised the following route for the efficient synthesis of **3**, a precursor of **4**. The sequence involved in the present synthesis consists of the diazotisation of **1** using hydrochloric acid and sodium nitrite and coupling with selected aromatic amino coupling components **2** such as *p*-toluidine **2a**, *p*-anisidine **2b**, *p*-chloroaniline **2c** to give excellent yields of 76-88% *o*-aminoaryl azo compounds **3a-3c**. The compounds **3a-3c** were converted to 7-triazolyl-1,4-dioxino[2,3-*b*]quinoxalines **4a-4c** using cupric acetate, in refluxing *N,N*-dimethylformamide, in a current of bubbling air [29].

As a part of our study to enhance the fluorescent characteristics of the heterocycle, it was planned to introduce one six membered ring and other five membered ring fused to quinoxaline at 2,3- and 6,7-positions, respectively and study the fluorescent properties of fused quinoxalines. Diazotised aryl amines such as aniline **5a**, *p*-toluidine **5b**, *p*-anisidine **5c** were coupled with **1** to give *o*-aminoaryl azo compounds **6a-6c**, which were oxidized with cupric acetate in DMF, in a current of bubbling air [29] afforded 1,4-dioxino[2,3-*b*]quinoxalino[6,5-*d*]-1,2,3-triazoles **7a-7c**.

The object of this present study was the synthesis of fused heterocyclic compounds. It was therefore planned to develop two six membered rings condensed to quinoxaline at 2,3- and 6,7- positions. It was envisaged to develop most compact structure of the heterocycles. Because of the compactness of structures of heterocyclic systems and flow of electrons from one part to the other resulted in enhancing the fluorescent properties of the compounds. 7-Amino-1,4-dioxino[2,3-*b*]quinoxaline **1** has been condensed with conjugated enol ethers such as diethyl ethoxymethylene malonate (EMME) **8a** and ethyl ethoxymethylenecyanoacetate (EMCA) **8b** afforded ethyl 7-aminoarylates **9a-9b** which were cyclised in dowtherm to

yield 1,4-dioxino[2,3-*b*]quinoxalino[6,5-*b*]pyridine **10a-10b**.

The fluorescent properties of the compounds **4a-4c**, **7a-7c** and **10a-10b** have been studied and the wavelength of absorption maxima, fluorescence emission maxima and the values of the logarithms of the extinction coefficients were recorded. The application to synthetic fibres (polyester) resulted in moderate whitening of the fibres. The compounds **4a-4c**, **7a-7c** and **10a-10b** possessed bluish-green fluorescence in daylight in most of the organic solvents.

EXPERIMENTAL

All melting points are uncorrected and are in 0°C. The infrared spectra were recorded on Perkin-Elmer Model 397 spectrophotometer in Nujol mull. The ¹H nmr spectra were recorded on Varian-60 MHz instrument EM-360-L using TMS as internal standard and the chemical shifts are given in δ (ppm) scale. Mass spectra were recorded on a Varian Mat-311 instrument (70 eV). Absorption and fluorescence emission spectra in DMF solution were recorded on Beckman Model-25 spectrophotometer and Aminco Spectrophotofluorometer, respectively.

7-(2-Amino-5-methylphenyl)azo-1,4-dioxino[2,3-*b*]quinoxaline (**3a**).

To a solution of 20 ml of concentrated hydrochloric acid, was dissolved 1.015 g (0.005 mole) of 7-amino-1,4-dioxino[2,3-*b*]quinoxaline **1** by warming and the solution was then cooled to 0-5°. With vigorous stirring, 0.38 g (0.0055 mole) of sodium nitrite in 3 ml water was gradually added to this solution in 1 hour at 0-5°. The reaction mixture was stirred for a further 1 hour, maintaining the temperature at 0-5°. The excess of nitrous acid was decomposed by the addition of urea. The clear diazonium salt solution was slowly added to 0.535 g (0.005 mole) of *p*-toluidine **2a** in 5 ml of acetic acid at 0°. The pH of the reaction mixture was maintained at 4-5 throughout the coupling period by addition of sodium carbonate in portions for 2 hours at 0°. After the addition of the diazonium salt was over, the reaction mixture was stirred for a further period of 5 hours and the partially separated dye was completely precipitated by neutralization. The dye was filtered, washed with water and dried. Recrystallisation from methanol yielded 1.41 g (88%) of **3a**, mp 211-212°.

Anal. Calcd. for C₁₇H₁₅N₅O₂: C, 63.55; H, 4.67; N, 21.80. Found: C, 63.51; H, 4.69; N, 21.83.

7-(2-Amino-5-methoxyphenyl)azo-1,4-dioxino[2,3-*b*]quinoxaline (**3b**).

The same procedure as described for **3a** was applied except *p*-anisidine **2b** was used in place of **2a**, yielding 7-(2-amino-5-methoxyphenyl)azo-1,4-dioxino[2,3-*b*]quinoxaline **3b**, recrystallised from methanol to yield 1.38 g (82%) of **3b**, mp 279-282°.

Anal. Calcd. for C₁₇H₁₅N₅O₃: C, 60.53; H, 4.45; N, 20.77. Found: C, 60.50; H, 4.49; N, 20.71.

7-(2-Amino-5-chlorophenyl)azo-1,4-dioxino[2,3-*b*]quinoxaline (**3c**).

The same procedure as described for **3a** was applied except *p*-chloroaniline **2c** was used in place of **2a**, yielding 7-(2-amino-5-chlorophenyl)azo-1,4-dioxino[2,3-*b*]quinoxaline **3c**, recrystallised from ethyl acetate to yield 1.29 g (76%) of **3c**, mp 237°.

Anal. Calcd. for $C_{16}H_{12}ClN_5O_2$: C, 56.30; H, 3.51; Cl, 10.26; N, 20.52. Found: C, 56.34; H, 3.53; Cl, 10.21; N, 20.50.

7-(5-Methylbenzo[1,2-*d*]-1,2,3-triazol-2-yl)-1,4-dioxino[2,3-*b*]-quinoxaline (**4a**).

To a solution of 1.605 g (0.005 mole) of **3a** in 15 ml of *N,N*-dimethylformamide was added 1 g (0.0052 mole) of cupric acetate. The reaction mixture was brought to reflux temperature. Air was continuously bubbled through the reaction mixture and the reflux was continued for 5 hours. The reaction mixture was cooled and then added to 100 ml of 5% ice cold hydrochloric acid with constant stirring. The precipitated solid was filtered, washed with water, dried and recrystallised from DMF to yield 1.13 g (71%) of **4a**, mp 328°; ¹H nmr (dimethyl sulfoxide-*d*₆): δ 2.8 (s, 3H, CH₃), 8.07-8.45 (m, 6H, aromatic), 3.75 (bs, 4H, aliphatic H-2, H-3); ms: *m/z* 319 (M⁺); λ max absorption 373 nm, λ max emission 436 nm, log ε 4.19.

Anal. Calcd. for $C_{17}H_{13}N_5O_2$: C, 63.94; H, 4.07; N, 21.94. Found: C, 63.91; H, 4.02; N, 21.96.

7-(5-Methoxybenzo[1,2-*d*]-1,2,3-triazol-2-yl)-1,4-dioxino[2,3-*b*]-quinoxaline (**4b**).

The same procedure as described for the oxidation of **3a** to **4a** was applied to the oxidation of **3b**, yielding 7-(5-methoxybenzo[1,2-*d*]-1,2,3-triazol-2-yl)-1,4-dioxino[2,3-*b*]-quinoxaline **4b**, recrystallised from DMF to yield 1.03 g (62%) of **4b**, mp 347-349°; ¹H nmr (dimethyl sulfoxide-*d*₆): δ 3.95 (s, 3H, OCH₃), 8.1-8.53 (m, 6H, aromatic), 3.79 (bs, 4H, aliphatic H-2, H-3); ms: *m/z* 335 (M⁺); λ max absorption 378 nm, λ max emission 440 nm, log ε 4.24.

Anal. Calcd. for $C_{17}H_{13}N_5O_3$: C, 60.89; H, 3.88; N, 20.89. Found: C, 60.91; H, 3.84; N, 20.82.

7-(5-chlorobenzo[1,2-*d*]-1,2,3-triazol-2-yl)-1,4-dioxino[2,3-*b*]-quinoxaline (**4c**).

The same procedure as described for the oxidation of **3a** to **4a** was applied to the oxidation of **3c** yielding 7-(5-chlorobenzo[1,2-*d*]-1,2,3-triazol-2-yl)-1,4-dioxino[2,3-*b*]-quinoxaline **4c**, recrystallised from DMF to yield 1.15 g (68%) of **4c**, mp 319°; ¹H nmr (dimethyl sulfoxide-*d*₆): δ 8.15-8.61 (m, 6H, aromatic), 3.71 (bs, 4H, aliphatic, H-2, H-3); ms: *m/z* 339 (M⁺); λ max absorption 382 nm, λ max emission 447 nm, log ε 4.31.

Anal. Calcd. for $C_{16}H_{10}ClN_5O_2$: C, 56.63; H, 2.94; Cl, 10.32; N, 20.64. Found: C, 56.59; H, 2.90; Cl, 10.28; N, 20.67.

7-Amino-6-(phenyl)azo-1,4-dioxino[2,3-*b*]-quinoxaline (**6a**).

To a solution of 10 ml of concentrated hydrochloric acid, was dissolved 0.46 g (0.005 mole) of aniline **5a** and the solution was then cooled to 0-5°. With vigorous stirring 0.38 g (0.0055 mole) of sodium nitrite in 3 ml water was gradually added to this solution in 1 hour at 0-5°. The reaction mixture was stirred for a further 1 hour, maintaining the temperature at 0-5°. The excess of nitrous acid was decomposed by the addition of urea. The clear diazonium solution was slowly added to 1.015 g (0.005 mole) of 7-amino-1,4-dioxino[2,3-*b*]-quinoxaline **1** in 5 ml of acetic acid at 0°. The pH of the reaction mixture was maintained at 4-5 throughout the coupling period by the addition of sodium carbonate in portions for 1 hour at 0°. After the addition of the diazonium salt was over the reaction mixture was stirred for a further period of 4 hours and the partially separated dye was completely precipitated by neutralization. The dye was filtered, washed with water and dried. Recrystallisation from benzene yielded 1.30 g (85%) of **6a**, mp 309-311°.

Anal. Calcd. for $C_{16}H_{13}N_5O_2$: C, 62.54; H, 4.23; N, 22.80. Found: C, 62.51; H, 4.27; N, 22.85.

7-Amino-6-(4-methylphenyl)azo-1,4-dioxino[2,3-*b*]-quinoxaline (**6b**).

The same procedure as described for **6a** was applied except *p*-toluidine **5b** was used in place of **5a**, yielding 7-amino-6-(4-methylphenyl)azo-1,4-dioxino[2,3-*b*]-quinoxaline **6b**, recrystallised from benzene to yield 1.30 g (81%) of **6b**, mp 261°.

Anal. Calcd. for $C_{17}H_{15}N_5O_2$: C, 63.55; H, 4.67; N, 21.80. Found: C, 63.51; H, 4.63; N, 21.82.

7-Amino-6-(4-methoxyphenyl)azo-1,4-dioxino[2,3-*b*]-quinoxaline (**6c**).

The same procedure as described for **6a** was applied except *p*-anisidine **5c** was used in place of **5a**, yielding 7-amino-6-(4-methoxyphenyl)azo-1,4-dioxino[2,3-*b*]-quinoxaline **6c**, recrystallised from ethyl acetate to yield 1.29 g (77%) of **6c**, mp 247-249°.

Anal. Calcd. for $C_{17}H_{15}N_5O_3$: C, 60.53; H, 4.45; N, 20.77. Found: C, 60.57; H, 4.47; N, 20.73.

2-*N*-(phenyl)-1,4-dioxino[2,3-*b*]-quinoxalino[6,5-*d*]-1,2,3-triazole (**7a**).

To a solution of 1.535 g (0.005 mole) of **6a** in 15 ml of *N,N*-dimethylformamide was added 1 g (0.0052 mole) of cupric acetate. The reaction mixture was brought to reflux temperature. Air was continuously bubbled through the reaction mixture and the reflux was continued for 5 hours. The reaction mixture was cooled and then added to 100 ml of 5% ice cold hydrochloric acid with constant stirring. The precipitated solid was filtered, washed with water, dried and recrystallised from DMF to yield 1.05 g (69%) of **7a**, mp 311-312°; ¹H nmr (dimethyl sulfoxide-*d*₆): δ 7.8-8.4 (m, 7H, aromatic), 3.7 (bs, 4H, aliphatic, H-2, H-3); ms: *m/z* 305 (M⁺); λ max absorption 385 nm, λ max emission 445 nm, log ε 4.27.

Anal. Calcd. for $C_{16}H_{11}N_5O_2$: C, 62.95; H, 3.60; N, 22.95. Found: C, 62.92; H, 3.64; N, 22.91.

2-*N*-(4-Methylphenyl)-1,4-dioxino[2,3-*b*]-quinoxalino[6,5-*d*]-1,2,3-triazole (**7b**).

The same procedure as described for the oxidation of **6a** to **7a** was applied to the oxidation of **6b**, yielding 2-*N*-(4-methylphenyl)-1,4-dioxino[2,3-*b*]-quinoxalino[6,5-*d*]-1,2,3-triazole **7b**, recrystallised from DMF to yield 1.14 g (72%) of **7b**, mp 339°; ¹H nmr (dimethyl sulfoxide-*d*₆): δ 2.9 (s, 3H, CH₃), 7.75-8.6 (m, 6H, aromatic), 3.75 (bs, 4H, aliphatic, H-2, H-3); ms: *m/z* 319 (M⁺); λ max absorption 388 nm, λ max emission 449 nm, log ε 4.32.

Anal. Calcd. for $C_{17}H_{13}N_5O_2$: C, 63.94; H, 4.07; N, 21.94. Found: C, 63.97; H, 4.09; N, 21.98.

2-*N*-(4-Methoxyphenyl)-1,4-dioxino[2,3-*b*]-quinoxalino[6,5-*d*]-1,2,3-triazole (**7c**).

The same procedure as described for the oxidation of **6a** to **7a** was applied to the oxidation of **6c**, yielding 2-*N*-(4-methoxyphenyl)-1,4-dioxino[2,3-*b*]-quinoxalino[6,5-*d*]-1,2,3-triazole **7c**, recrystallised from DMF to yield 1.05 (63%) of **7c**, mp > 360°; ¹H nmr (dimethyl sulfoxide-*d*₆): δ 4.1 (s, 3H, OCH₃), 7.9-8.5 (m, 6H, aromatic), 3.77 (bs, 4H, aliphatic, H-2, H-3); ms: *m/z* 335 (M⁺); λ max absorption 390 nm, λ max emission 453 nm, log ε 4.38.

Anal. Calcd. for $C_{17}H_{13}N_5O_3$: C, 60.89; H, 3.88; N, 20.89. Found: C, 60.92; H, 3.84; N, 20.93.

Ethyl 1,4-Dioxino[2,3-*b*]quinoxaline-7-(β -carbethoxyaminoacrylate) (**9a**).

A mixture of 1.015 g (0.005 mole) of 7-amino-1,4-dioxino[2,3-*b*]quinoxaline **1** and (0.005 mole) of ethyl ethoxymethylenemalonate ester (EMME) **8a** was heated in oil bath at 160° for 2 hours. The reaction mixture was cooled to room temperature and slowly added over ice-water mixture with constant stirring. The precipitated solid was filtered, washed with water, dried and recrystallised from ethanol to yield 1.26 g (68%) of **9a**, mp 234-235°.

Anal. Calcd. for C₁₈H₁₉N₃O₆: C, 57.90; H, 5.09; N, 11.26. Found: C, 57.88; H, 5.03; N, 11.31.

Ethyl 1,4-Dioxino[2,3-*b*]quinoxaline-7-(β -cyanoaminoacrylate) (**9b**).

The same procedure as described for **9a** was applied except ethyl ethoxymethylenecyanoacetate (EMCA) **8b** was used in place of **8a**, yielding ethyl 1,4-dioxino[2,3-*b*]quinoxaline-7-(β -cyanoaminoacrylate) **9b**, recrystallised from ethanol to yield 0.99 g (61%) of **9b**, mp 316-318°.

Anal. Calcd. for C₁₆H₁₄N₄O₄: C, 58.89; H, 4.29; N, 17.17. Found: C, 58.94; H, 4.26; N, 17.13.

3-Carbethoxy-4-hydroxy-1,4-dioxino[2,3-*b*]quinoxalino[6,5-*b*]pyridine (**10a**).

A mixture of 1.865 g (0.005 mole) of **9a** and 10 ml of dowertherm was refluxed for 6 hours. The reaction mixture was then cooled and diluted with 50 ml of petroleum ether. The precipitated solid was filtered, washed with petroleum ether and dried. Recrystallisation from DMF yielded 1.25 g (77%) of **10a**, mp > 360°; ir (nujol): 1700, 3500 cm⁻¹; ¹H nmr (dimethyl sulfoxide-d₆): δ 8.1-8.46 (m, 3H, aromatic), 3.75 (bs, 4H, aliphatic, H-2, H-3), 4.2 (q, 2H, CH₂), 1.4 (t, 3H, CH₃), 11.4 (s, 1H, OH); ms: m/z 327 (M⁺); λ max absorption 393 nm, λ max emission 462 nm, log ϵ 4.41.

Anal. Calcd. for C₁₆H₁₃N₃O₅: C, 58.71; H, 3.97; N, 12.84. Found: C, 58.73; H, 3.92; N, 12.81.

3-Cyano-4-hydroxy-1,4-dioxino[2,3-*b*]quinoxalino[6,5-*b*]pyridine (**10b**).

The same procedure as described for the cyclisation of **9a** to **10a** was applied for the cyclisation of **9b**, yielding 3-cyano-4-hydroxy-1,4-dioxino[2,3-*b*]quinoxalino[6,5-*b*]pyridine **10b**, recrystallised from DMF to yield 1.02 g (73%) of **10b**, mp > 360°; ir (nujol): 2220, 3400 cm⁻¹; ¹H nmr (dimethyl sulfoxide-d₆): δ 8.02-8.3 (m, 3H, aromatic), 3.78 (bs, 4H, aliphatic, H-2, H-3), 11.5 (s, 1H, OH); ms: m/z 280 (M⁺); λ max absorption 397 nm, λ max emission 465 nm, log ϵ 4.47.

Anal. Calcd. for C₁₄H₈N₄O₃: C, 60.00; H, 2.85; N, 20.00. Found: C, 60.03; H, 2.81; N, 20.06.

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